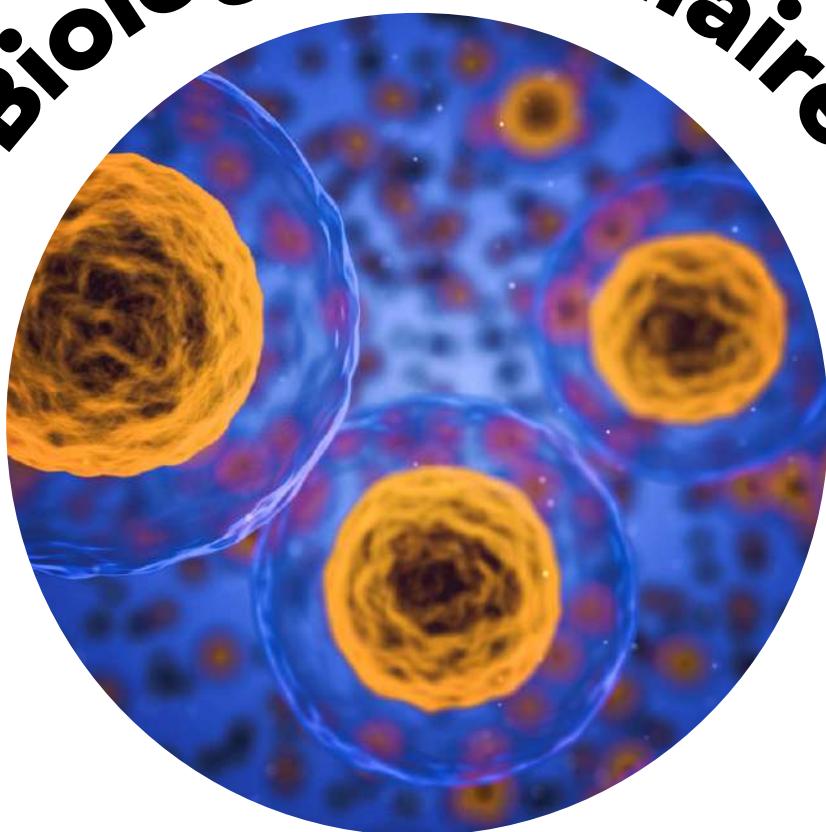


# Biologie Cellulaire



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VIE ET DE LA TERRE



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Etudier



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Emploi



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# Krebs cycle

Josef Fontana

EC - 46



# Overview of the seminar

- Krebs cycle (KC)
  - The importance of the KC for the cell - amphibolic character
  - The overall equation and the individual reactions of the KC
  - Anaplerotic reactions of the KC and the linking of the KC to with the other metabolic pathways in the cells
  - Regulation of the KC



## Sir Hans Adolf Krebs (25.8.1900 - 22.11.1981)

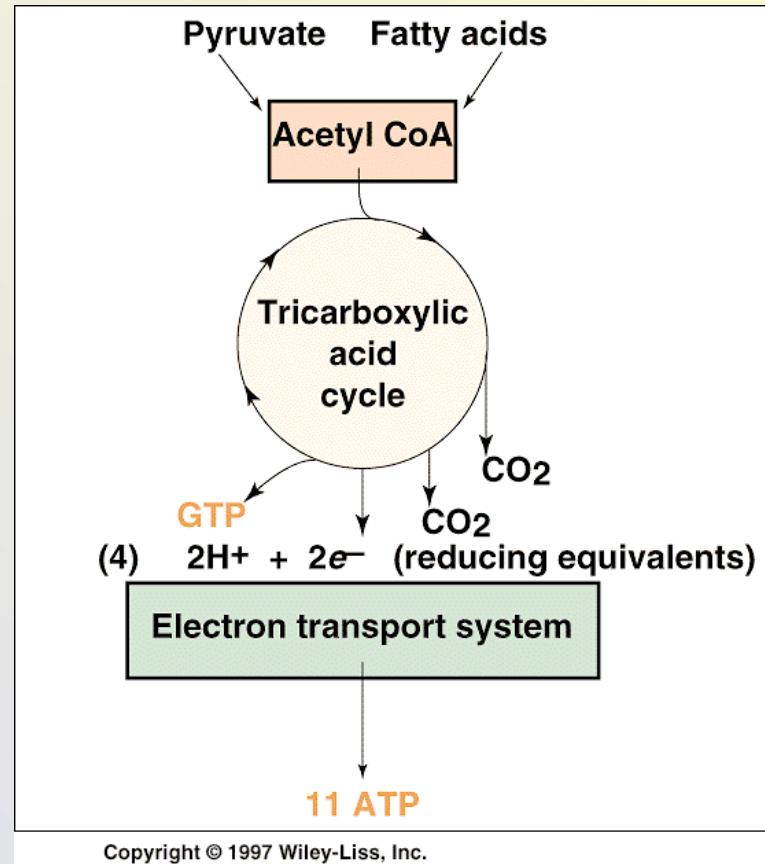
Was a German-born British physician and biochemist. The Nobel Prize in Physiology or Medicine in 1953 for his discovery of the citric acid cycle.

# Krebs cycle

The importance of the KC for  
the cell - amphibolic  
character

# Citric acid cycle (CAC)

- CAC takes place in the MIT matrix (not in erythrocytes) and only under aerobic conditions.
- It is the ultimate path of oxidation.
- Substrate is **Acetyl~CoA**, which is oxidised to  $2\text{CO}_2$  with simultaneous reduction of cofactors ( $3\text{NAD}^+$  and  $1\text{FAD}$ ). They are then reoxidised in RC.
- We get  $1\text{GTP}$  directly (= ATP).
- CAC plays a key role in further metabolic reactions (i. e. gluconeogenesis, transamination, deamination or lipogenesis).

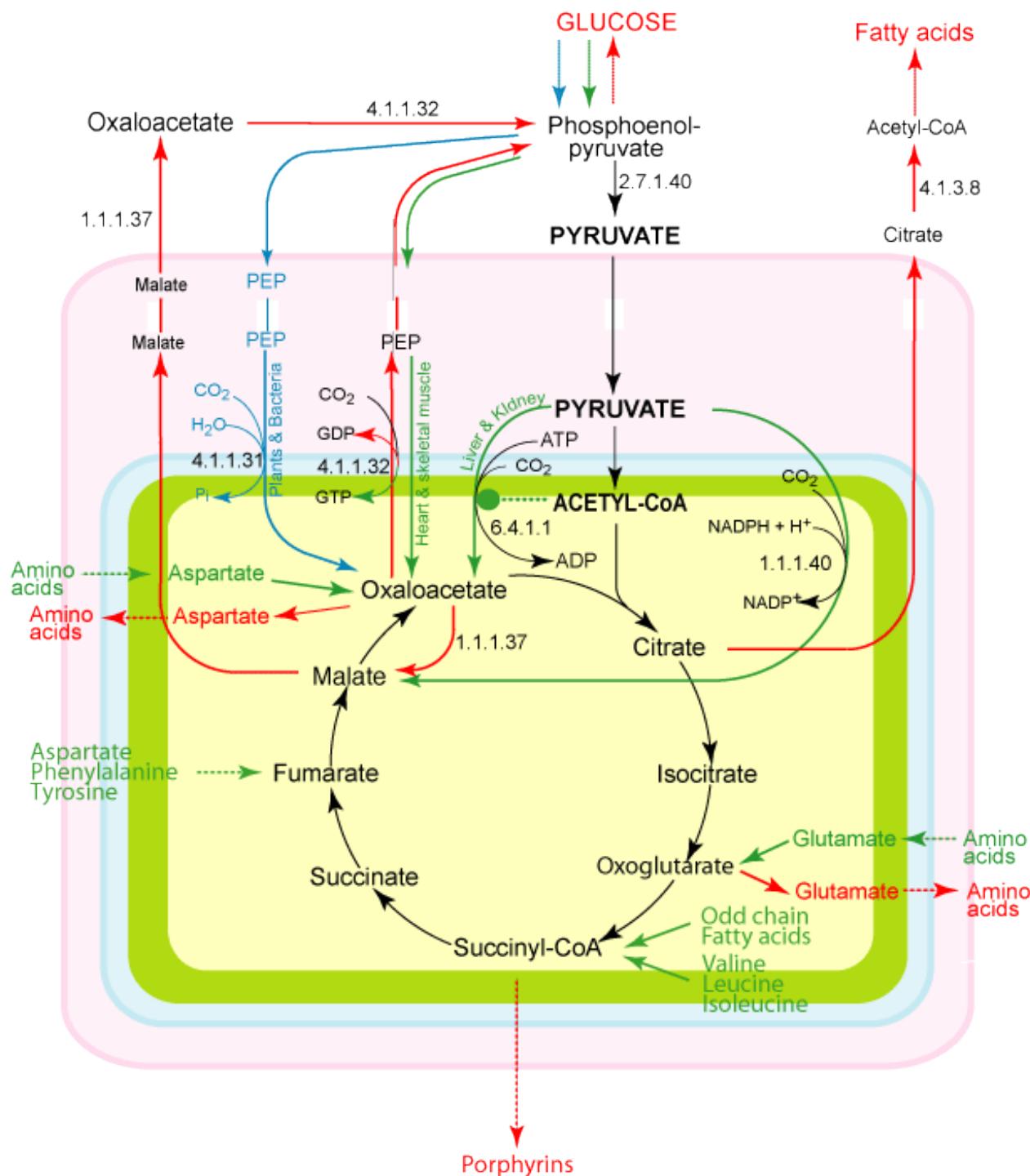


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The figure is adopted from the book: Devlin, T. M. (editor): Textbook of Biochemistry with Clinical Correlations, 4th ed. Wiley-Liss, Inc., New York, 1997. ISBN 0-471-15451-2

# Amphibolic character

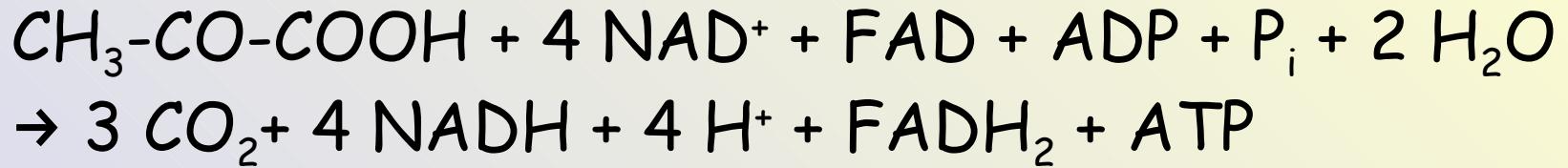
- It is not only catabolic pathway, but its intermediates create many biologically important compounds.
- E.g.:
  - From  $\alpha$ -ketoglutarate via transamination → AA glutamate, which is the most abundant excitatory neurotransmitter in the brain.
  - From Suc ~ CoA → tetrapyrroles (Heme).
- Therefore has KC also amphibolic function.



# Krebs cycle

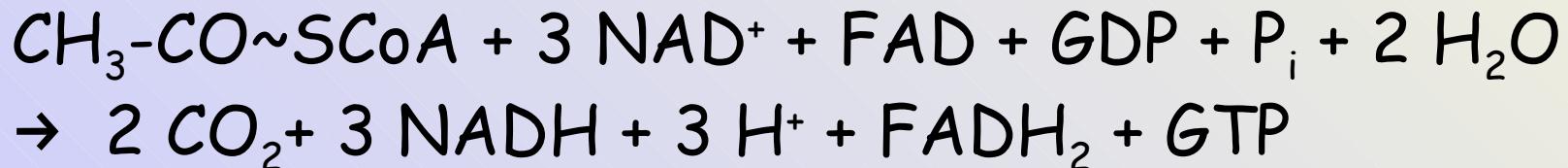
The overall equation and the  
individual reactions of the  
KC

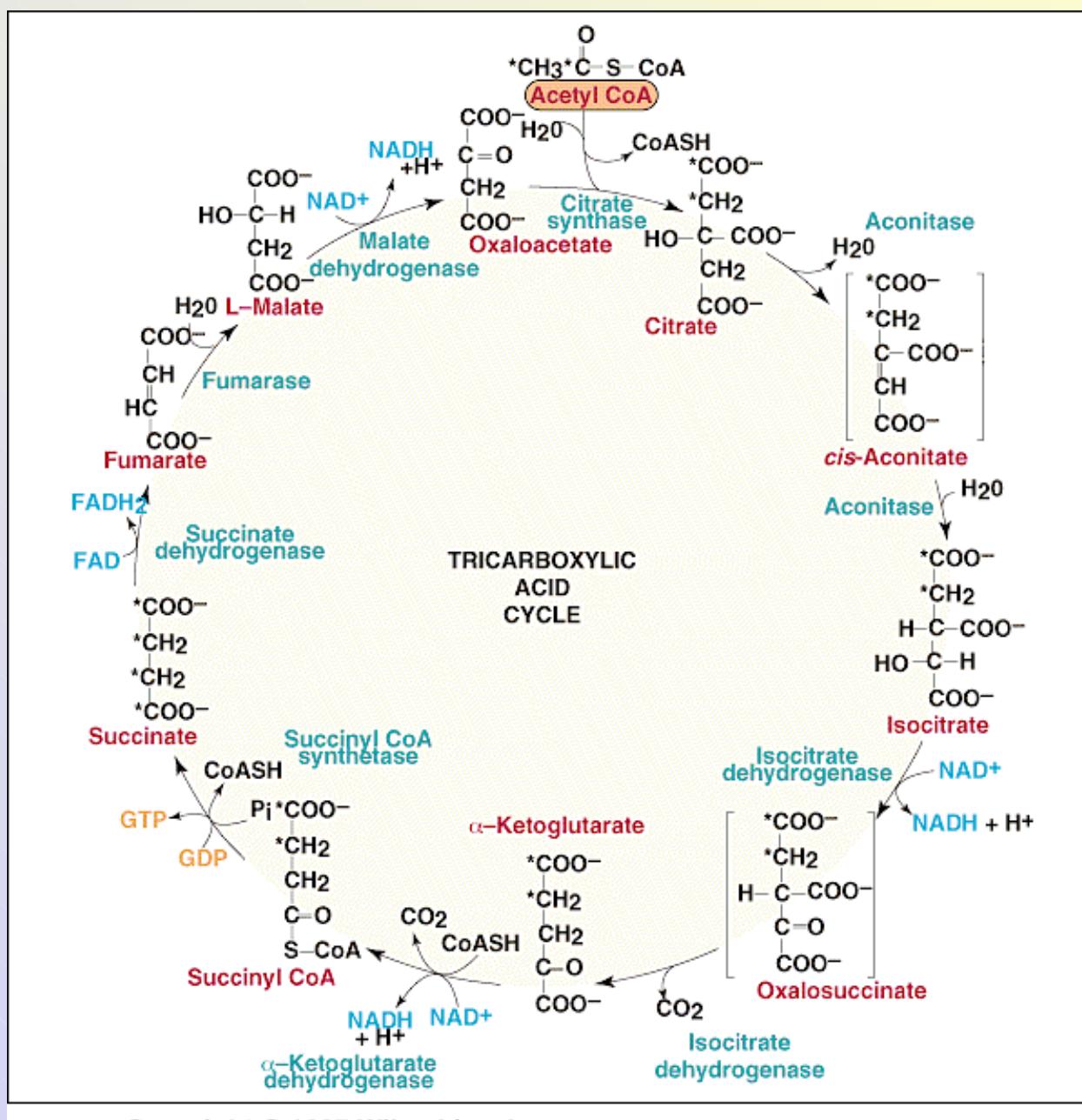
# Overall equation



- Pyruvate dehydrogenase reaction (oxidative decarboxylation, mitochondrial matrix)  
Irreversible reaction!
- $\text{CH}_3\text{-CO-COOH} + \text{NAD}^+ + \text{HSCoA} \rightarrow \text{CO}_2 + \text{NADH} + \text{H}^+ + \text{CH}_3\text{-CO}\sim\text{SCoA}$

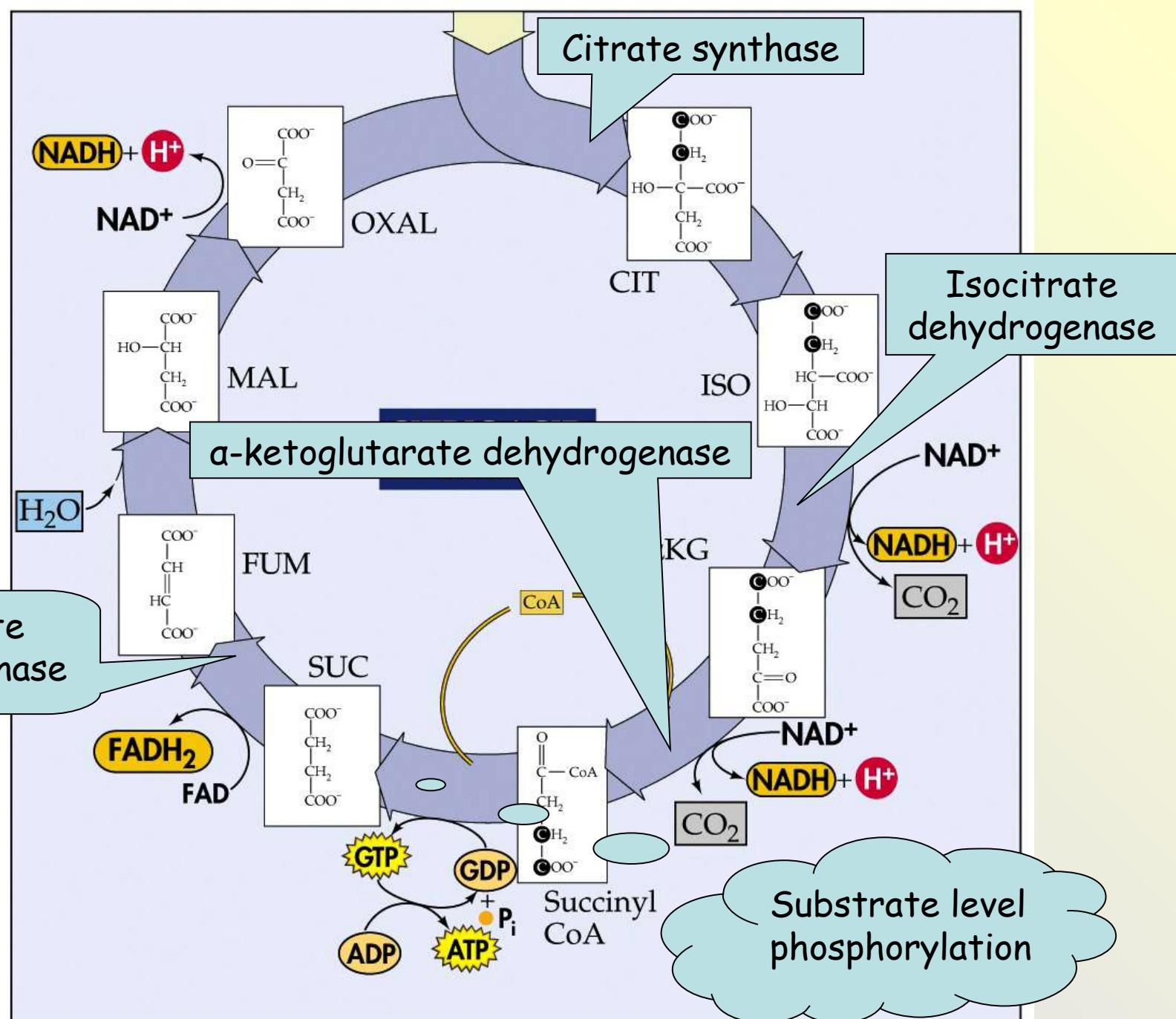
## Krebs cycle





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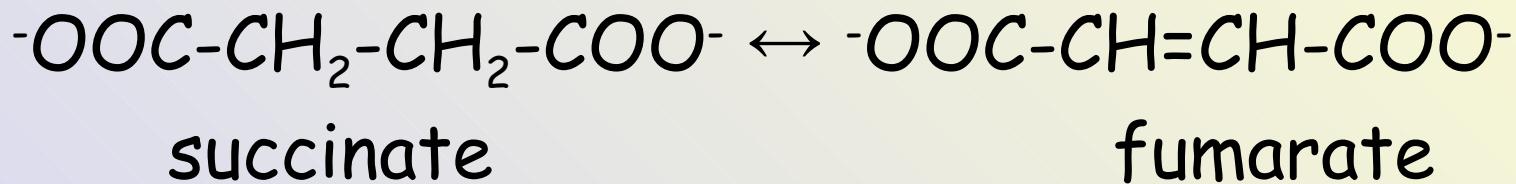


# Substrate level phosphorylation in KC

- In the KC, there is one substrate level - thioester bond in  $\text{Suc}\sim\text{CoA}$  decays with water to form succinate and  $\text{HSCoA}$ .
- This reaction releases so much energy that it can be coupled with the phosphorylation of GDP to GTP (exchanged for ATP).
- This reaction may also be bypassed in the degradation of ketone bodies, acetoacetate reacts with the  $\text{Suc}\sim\text{CoA}$  to form succinate and acetoacetyl $\sim\text{CoA}$ . But we gain no GTP.

# Succinate dehydrogenase

- In the KC it catalyzes:



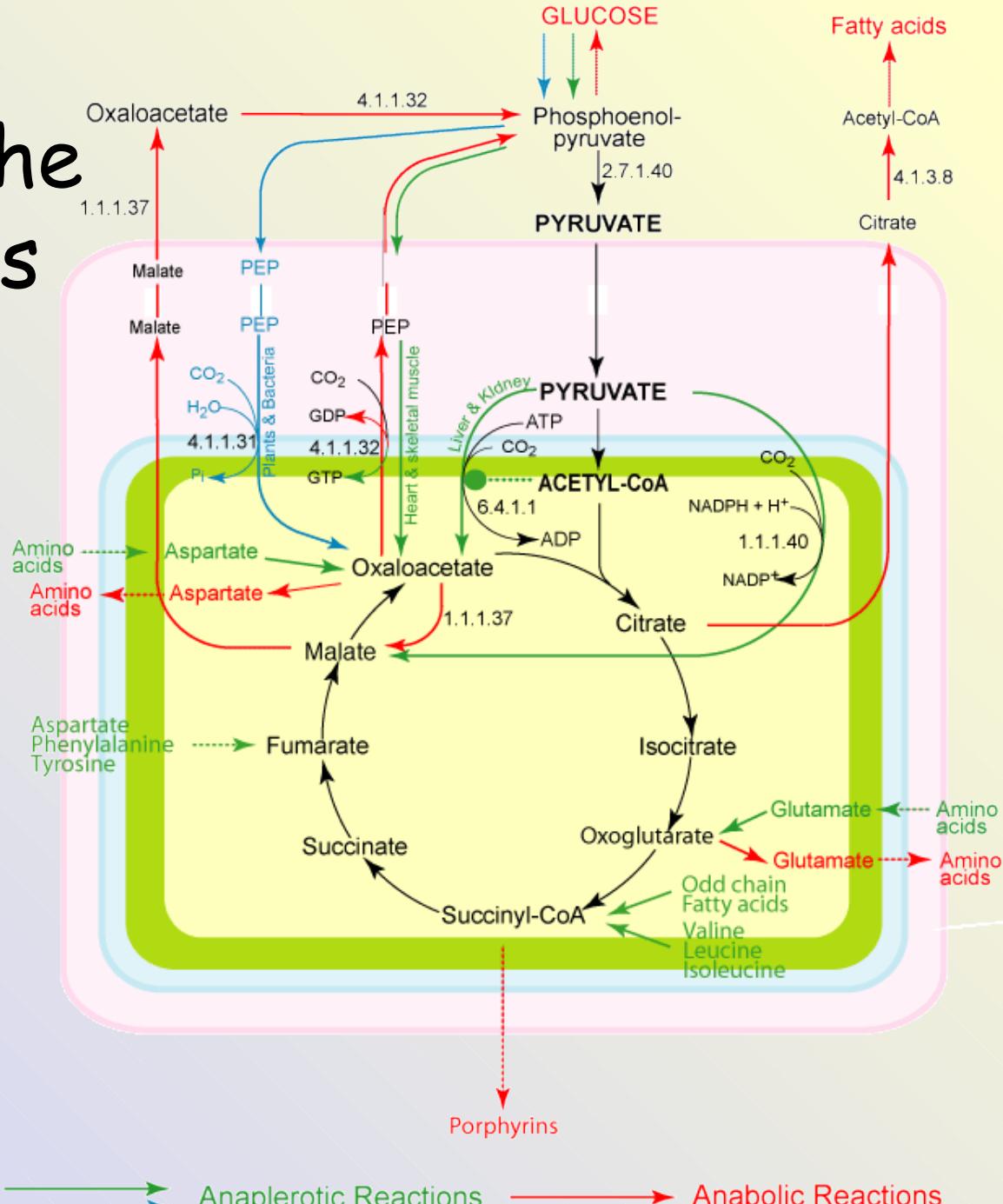
- Its prosthetic group is FAD.
- But it is also complex II in the inner mitochondrial membrane (respiratory chain). Electrons from  $\text{FADH}_2$  are passed directly to complex III.

# Krebs cycle

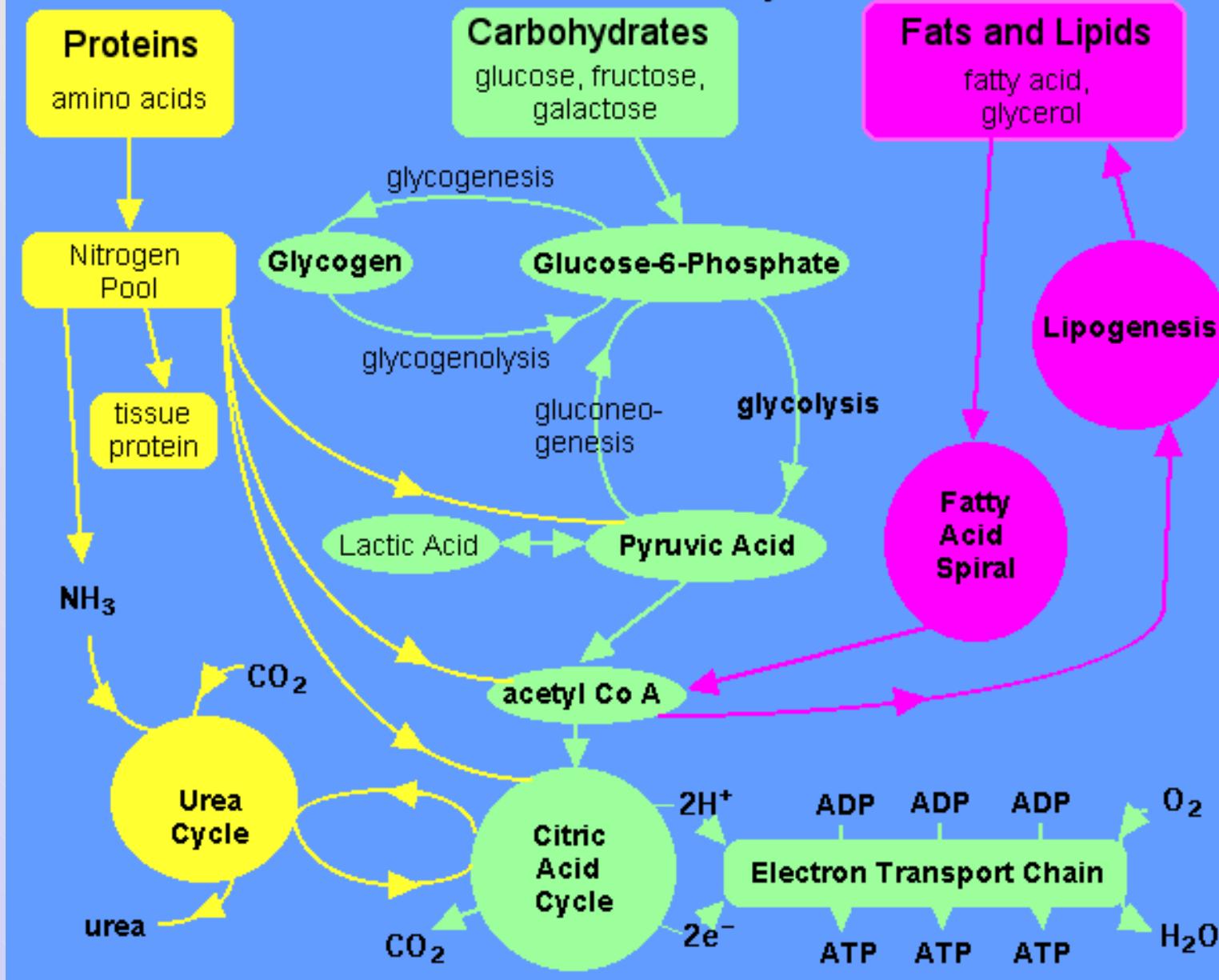
Anaplerotic reactions of the KC  
and the linking of the KC to  
with the other metabolic  
pathways in the cells

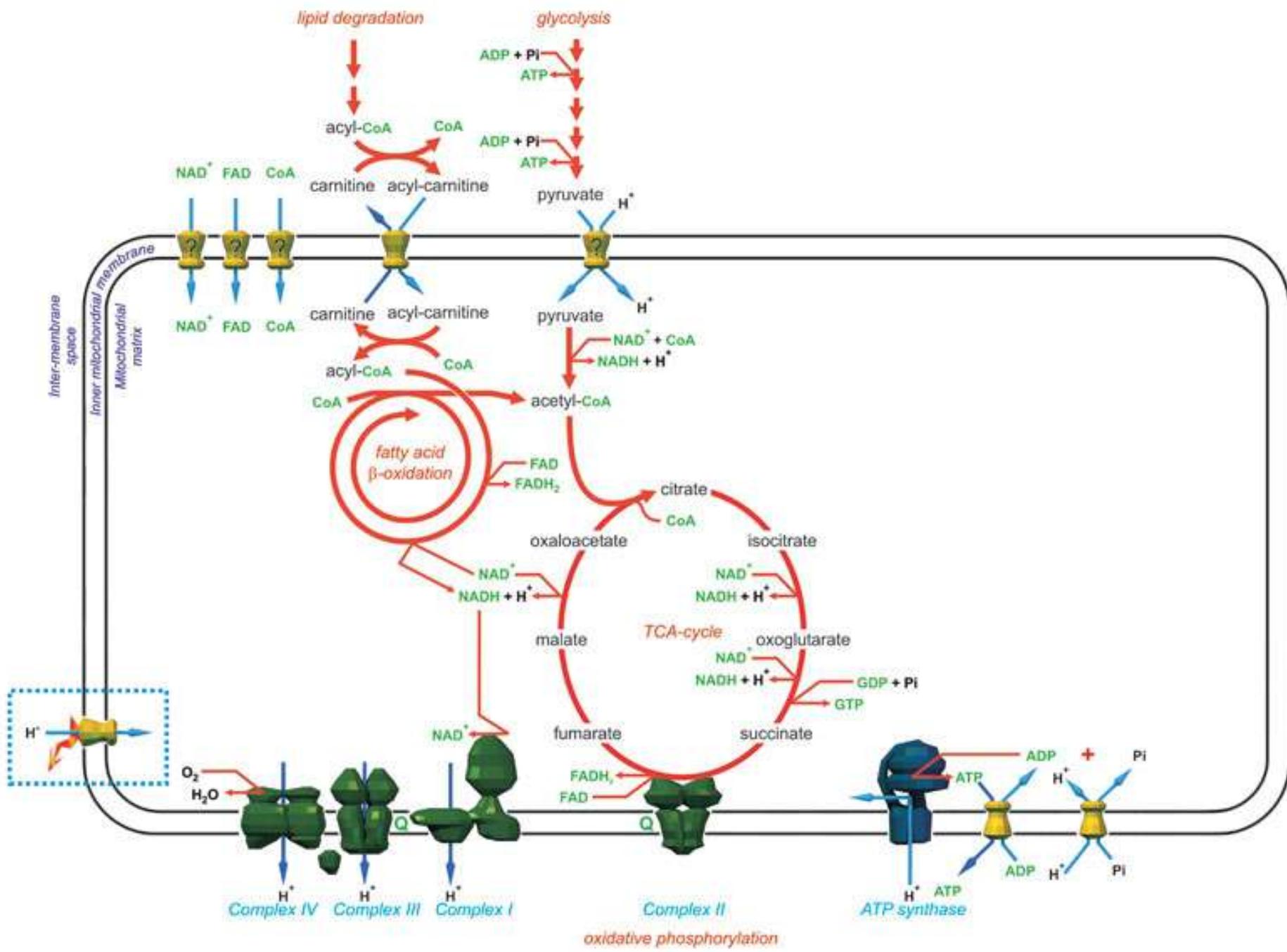
# Connection to the other pathways

- Respiratory chain
- Metabolism of amino acids
- Urea cycle
- Synthesis of heme
- Gluconeogenesis
- FA synthesis
- Citrate cycle as a source of substrates used in a synthesis of other molecules.



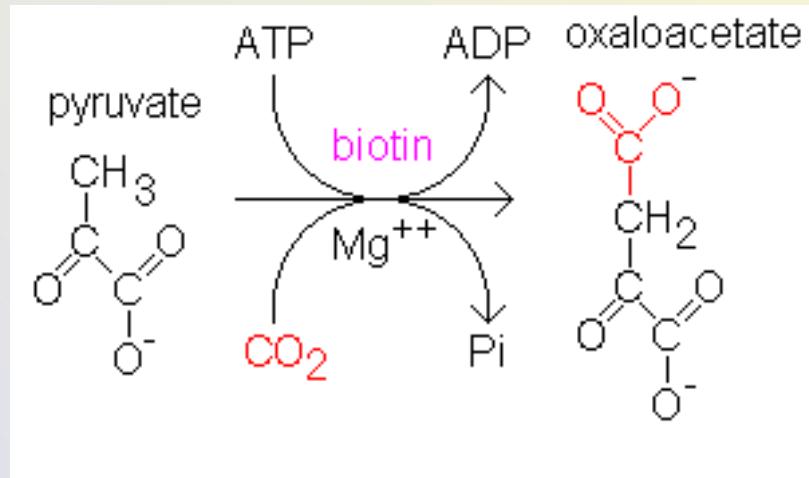
## Metabolism Summary





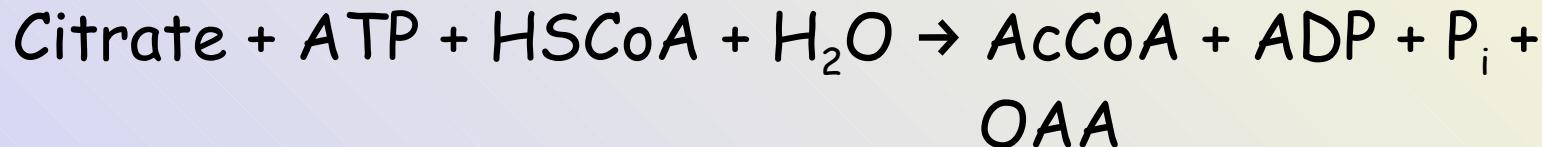
# Anaplerotic (support) reactions

- Because KC intermediates serve as substrates for other metabolic pathways, there are reactions to replace these losses.
- Synthesis of OAA from Pyr (Pyr carboxylation needs ATP) by pyruvate carboxylase.**
  - This is also the first step in gluconeogenesis.
  - Biotin is a cofactor of carboxylases.
- Degradation of most amino acids gives the following intermediates of KC: OAA,  $\alpha$ -KG, fumarate



# Connecting of KC to FA formation

- Acetyl-CoA + OAA → citrate (citrate synthase in KC)
- Citrate is transferred from mitochondria into the cytoplasm, (to MIT goes malate).
- Citrate in the cytoplasm is split into acetyl-CoA and OAA (ATP-citrate lyase).



- AcCoA enters the synthesis of FA.
- Reduction of oxaloacetate to malate (malate dehydrogenase = consumption of NADH + H +)
- Malate returns via antiport in mitochondria or is decarboxylated to pyruvate.

# Connecting of KC to FA formation

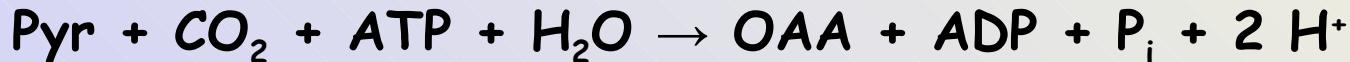
- OAA formed during transport of citrate into the cytosol should return back to the matrix. But the inner membrane of mitochondria is impermeable to OAA.
- OAA is therefore reduced in the presence of NADH to malate by the cytosolic malate dehydrogenase:



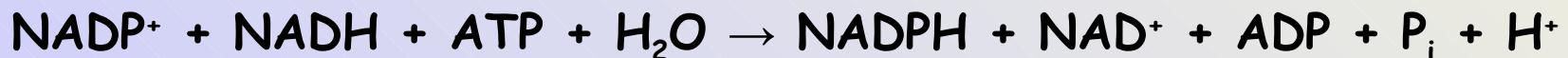
- Malate is oxidatively decarboxylated by NADP<sup>+</sup>- malate enzyme (malic enzyme):



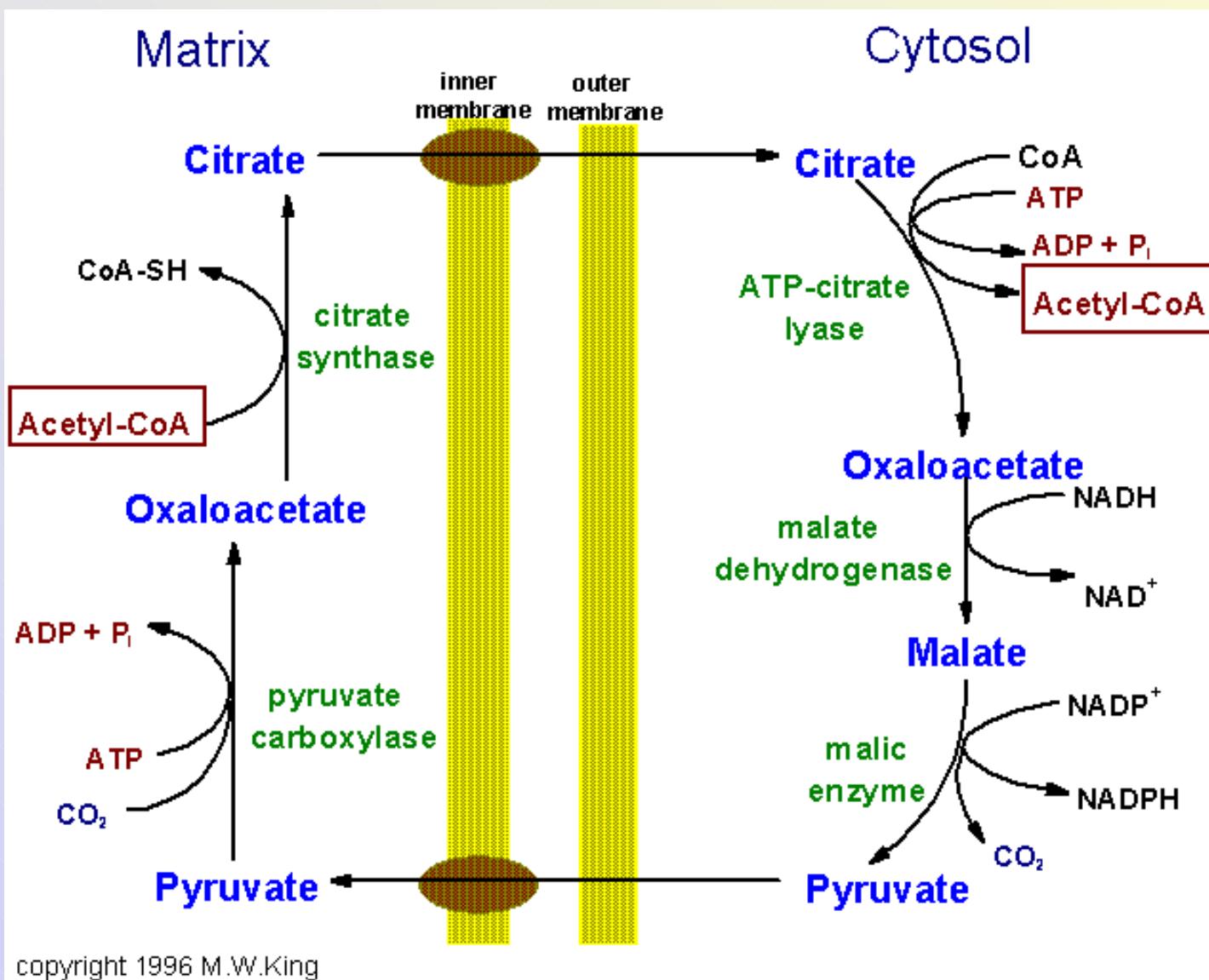
- Finally, Pyr enters the mitochondria, where it is carboxylated by pyruvate carboxylase:



- Summary equation:



# Connecting of KC to FA formation



# Krebs cycle

## Regulation of the KC

# Regulatory enzymes

- Citrate synthase
- Isocitrate dehydrogenase
- $\alpha$ -ketoglutarate dehydrogenase
- The activity of CAC is closely linked to the availability of  $O_2$ .

# Regulation of the KC

- Regulatory enzymes:
  - 1) Citrate synthase
  - 2) Izocitrate dehydrogenase
  - 3)  $\alpha$ -ketoglutarate dehydrogenase
- Energy Control: If enough energy, enzymes 2 and 3 are inhibited - ATP is inhibitor, ADP is activator.
- Respiratory Control: KC and RC must work together. Accumulation of NADH+H<sup>+</sup> and FADH<sub>2</sub> inhibits enzymes 2, 3.
- Substrate control: Enzyme 1. Citrate synthase is mainly regulated by the availability of acetyl-CoA and oxaloacetate.

# Bon courage



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